## BACHELOR 'S PROGRAMME 2<sup>nd</sup> YEAR OF STUDY, 1<sup>st</sup> SEMESTER

COURSE TITLE	GENERAL CHEMISTRY	
COURSE CODE		
COURSE TYPE	full attendance	
COURSE LEVEL	1 <sup>st</sup> cycle (bachelor's degree)	
YEAR OF STUDY, SEMESTER	2 <sup>nd</sup> year of study, 1 <sup>st</sup> semester	
NUMBER OF ECTS CREDITS	4	
NUMBER OF HOURS PER WEEK	4 (2 lecture hours + 2 seminar hours)	
NAME OF LECTURE HOLDER	Assoc.prof.PhD. Danut Cozma	
NAME OF SEMINAR HOLDER	Assoc.prof.PhD . Danut Cozma	
PREREQUISITES	Advanced level of English	
A GENERAL AND COURSE-SPE		
General competences:		
$\rightarrow$ Possession of co	$\rightarrow$ Possession of cognitive / cognitive transfer skills by analogy specific to the domain and field of	
specialization and	specialization and their proper use in a given professional context Possession of cognitive skills /	
techniques and m	ethods of learning / assessment applicable to the field and field of specialization	
and their use to id	entify their own training needs / training needs in a team-building type process.	
Course-specific competences:		
$\rightarrow$ Appropriate use of	the theoretical foundations of applied engineering sciences. Providing research	
support activities. Use of standard laboratory or industrial laboratory equipment for conducting		
research experime	ents	
B LEARNING OUTCOMES		
Upon successful completion of this discipline, students will be able to:		
elements.		
-Explates the correlation between the type of chemical bonds and the properties of the substances		
C LECTURE CONTENT		
Fundamental laws	of chemistry.	
Classification of elements. Study of periodic and non-periodic properties. Non-metallic function     and metallic function of the elements		
<ul> <li>Types of chemical honds. Correlation chemical honds-properties of substances</li> </ul>		
<ul> <li>Chemical reaction.</li> </ul>		
• Metals and non-metals - methods of obtaining and purifying. Compound substances: oxides,		
bases, acids, salts		
Metals and non-r     bases acids salts	netals - methods of obtaining and purifying. Compound substances: oxides,	
D RECOMMENDED READING FC	R I FCTURES	
1.Mirela Goantă, Ioana Au	relia Gorodea, Fundamentele chimiei, Ed. Stef. 2012.	
2.S.Ifrim, I.Rosca. Chimie generala. Ed.Tehnica, Bucuresti, 1989.		
3. Gh. Marcu, M. Rusu, V. Coman – Chimie anorganica. Semimetale si nemetale, Editura Eikon, 2007.		
4.D.Ganju, "Substanțe tehnice anorganice", Ed.Univ."Al.I.Cuza" Iași1997		
6. Industrial inorganic pigments I ed. by Gunter Buxbaum. Wilev-VCH. 1998.		
7. Gerald F. Dionne. Magnetic Oxides. Springer Science+Business Media, LLC 2009.		
8. Frank J. Owens, Charle	s P. Poole, JrThe New Superconductors. Kluwer Academic Publishers.2002.	
E SEMINAR CONTENT		
Processing of wor     Chamical aplaulat	< safety rules. Presentation of the laboratory theme	
<ul> <li>Solutions Modes of expression of solution concentrations. Determination of solubility of</li> </ul>		
substances. Determination of crystallization water from crystalline hydrides - CuSO4 * 5H2O.		
Methods for purification and separation of substances (filtration, recrystallization, sublimation).		
Physical and chemical transformations		
<ul> <li>Classification of chemical reactions. Chemical reactions in aqueous medium (proton transfer reactions, electron transfer reactions).</li> </ul>		
reactions, electron     Obtaining classes	of compounds of metals and non-metals	
F RECOMMENDED READING EC	R SEMINARS	
1. J. A. Beran, Laboratory Manual for Principles of General Chemistry, John Wiley & Sons 2011.		
2.Handbook of prepar	ative Inorganic Chemistry, Edited by G.Bauer A Press, London 1963	

3.Spencer L. Seager, Michael R. Slabaugh. Safety-Scale Laboratory Experiments for Chemistry for Today: General, Organic, and Biochemistry, 7e. Brooks/Cole, Cengage Learning. 2011.	
G EDUCATION STYLE	
LEARNING AND TEACHING METHODS	Lecture, debate, conversation, laboratory experiment, exercise and problem solving
ASSESSMENT METHODS	<ul><li>Written exam</li><li>Assessment along the way</li></ul>
LANGUAGE OF INSTRUCTION	English